# Synthesis of Pyrazolo[3,4-d]pyrimidines by Intramolecular Cycloaddition of Azahexatrienes 

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Treatment of aldehyde pyrimidin-6-ylhydrazones with aromatic aldehydes (or treatment of 6-hydrazinopyrimidines with aromatic aldehydes in excess) in dimethylformamide under reflux gave 2,3-disubstituted pyrazolo[3,4-d]pyrimidines.

Previous methods for the preparation of pyrazolo-[3,4-d]pyrimidines ${ }^{1}$ involve (a) construction of suitably substituted pyrazole precursors followed by pyrimidine ring closure, ${ }^{, 2,3}$ (b) condensation-cyclization of 5 substituted 6 -hydrazinopyrimidines, ${ }^{4,5}$ and (c) oxidative cyclization of 6 -benzylidenehydrazinopyrimidine derivatives. ${ }^{6}$ We now present the experimental details of a new approach to pyrazolo $[3,4-d]$ pyrimidines. ${ }^{7}$

Refluxing aldehyde 1,3 -dimethyluracil- 6 -ylhydrazones $(1 \mathrm{a}-\mathrm{g})^{6}$ with aromatic aldehydes in dimethylformamide gave the corresponding 2,3 -disubstituted 5,7 -dimethyl-pyrazolo[3,4-d]pyrimidine-4,6(5H,7H)-diones ( $2 \mathrm{a}-\mathrm{k}$ ). The structures (2) were confirmed by the benzylic proton signals at $\delta 5.3-5.5$ in the n.m.r. spectra (see Supplementary Publication) and by comparison with authentic samples prepared by benzylation of 3 -aryl-5,7-dimethylpyrazolo $[3,4-d]$ pyrimidine-4,6(5H,7H)diones ${ }^{6}$ with benzyl halides and potassium carbonate in dimethylformamide. $\dagger$
Heating the 6-benzylidenehydrazino-3-methyluracils $(3 a-c)^{6}$ with aromatic aldehydes in ethanol gave the 5 -benzylidene derivatives ( $4 \mathrm{a}-\mathrm{d}$ ) possessing a diaza-hexatriene-type structure. Refluxing the products (4ad) in dimethylformamide led to the 3 -aryl-2-benzyl- 5 methylpyrazolo $[3,4-d]$ pyrimidine-4,6( $5 \mathrm{FH}, 7 \mathrm{H})$-diones
( $5 \mathrm{a}-\mathrm{d}$ ). Refluxing of compounds ( $3 \mathrm{a}-\mathrm{c}$ ) with aromatic aldehydes in dimethylformamide gave the pyrazolopyrimidines ( $5 \mathrm{a}-\mathrm{d}$ ) directly; these products were identified by transformation into compounds (2) by methylation with methyl iodide and potassium carbonate in dimethylformamide.

When acetophenone 1,3 -dimethyluracil- 6 -ylhydrazone (6) ${ }^{8}$ was treated with benzaldehyde in dimethylformamide, 2 -(1-methylbenzyl)-5,7-dimethyl-3-phenylpyrazolo $3,4-d]$ pyrimidine-4,6( $5 \mathrm{H}, 7 \mathrm{H}$ )-dione (7) was obtained.

Heating the 2 -amino-6-benzylidenehydrazino-4-hydroxypyrimidines $(8 \mathrm{a}-\mathrm{d}){ }^{9}$ with aromatic aldehydes [or heating 2 -amino-6-hydrazino-4-hydroxypyrimidine ( 10$)^{9}$ with an excess of aromatic aldehyde] in dimethyl-

[^0]formamide gave the 6-amino-3-aryl-2-benzyl-4-hydr-oxypyrazolo[3,4-d]pyrimidines (9a-d).

## EXPERIMENTAL

2,3-Disubstituted 5,7-Dimethylpyrazolo[3,4-d]pyrimidine$4,6(5 \mathrm{H}, 7 \mathrm{H})$-diones $(2 \mathrm{a}-\mathrm{k})$. General Procedure.-Method $A$. To a solution of an aldehyde 1,3 -dimethyluracil-6ylhydrazone (1) ( 0.008 mol ) in dimethylformamide ( 50 ml ) was added an aromatic aldehyde ( 0.01 mol ). The mixture was refluxed for 7 h , then evaporated to dryness in vacuo, and the residue was recrystallized from ethanol to give the corresponding pyrazolo[3,4-d]pyrimidine ( $2 \mathrm{a}-\mathrm{k}$ ) (Table $1) . \ddagger$ M.p.s. and yields are indicated below the formulae.

Method B. A mixture of the pyrazolopyrimidine (5) (see later) ( 0.002 mol ), methyl iodide ( 0.01 mol ), and potassium carbonate ( 0.01 mol ) in dimethylformamide ( 20 ml ) was refluxed for 2 h . Inorganic substances were filtered off, the filtrate was evaporated to dryness in vacuo, and the residue was diluted with water. Recrystallization of the resulting crystals from ethanol gave the 5,7 -dimethylpyrazolo $[3,4-d]$ pyrimidinediones, which were identical with the products from method $A$. Compounds ( $2 \mathrm{a}-\mathrm{c}$ ) were obtained in 70, 74, and $64 \%$ yield, respectively.

5-Benzylidene-6-benzylidenehydrazino-3-methyluracils
(4a-d). General Procedure.-A mixture of a 6-benzyl-idenehydrazino-3-methyluracil (3) (0.008 mol) and an aromatic aldehyde ( 0.01 mol ) in ethanol ( $100-150 \mathrm{ml}$ ) was refluxed for 3 h . After cooling, the crystals which separated were filtered off and recrystallized from ethanol. The following compounds were obtained: 5-benzylidene-6-benzylidenehydrazino-3-methyluracil (4a) (88\%), m.p. $279^{\circ}$ (Found: C, 68.45; H, 4.9; N, 16.75. $\mathrm{C}_{19} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $\mathrm{C}, 68.65 ; \mathrm{H}, 4.85 ; \mathrm{N}, 16.85 \%$ ) ; 5-(p-chlorobenzylidene)-6-(p-chlorobenzylidenehydrazino)-3-methyluracil (4b) (93\%), m.p. $298^{\circ}$ (Found: C, 56.7 ; $\mathrm{H}, 3.7$; N, 14.0. $\mathrm{C}_{19} \mathrm{H}_{14} \mathrm{Cl}_{2}$ $\mathrm{N}_{4} \mathrm{O}_{2}$ requires C, $\left.56.85 ; \mathrm{H}, 3.5 ; \mathrm{N}, 13.95 \%\right)$; 5 -(p-methoxy-benzylidene)-6-(p-methoxybenzylidenehydrazino)-3-methyl-
uracil (4c), ( $59 \%$ ), m.p. $267^{\circ}$ (Found: C, 63.7; H, 5.55; $\mathrm{N}, 14.0 . \quad \mathrm{C}_{21} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{4}$ requires $\mathrm{C}, 63.95 ; \mathrm{H}, 5.6 ; \mathrm{N}, 14.2 \%$ ); 5-(p-chlorobenzylidene)-6-(p-methoxybenzylidenehydrazino)-
3 -methyluracil (4d) (72\%), m.p. 258 ${ }^{\circ}$ (Found: C, 60.8; $\mathrm{H}, 4.3 ; \mathrm{N}, 14.05 . \mathrm{C}_{20} \mathrm{H}_{17} \mathrm{ClN}_{4} \mathrm{O}_{3}$ requires $\mathrm{C}, 60.55 ; \mathrm{H}, 4.3$; $\mathrm{N}, 14.1 \%$ ).
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## 3-Aryl-2-benzyl-5-methylpyrazolo[3,4-d]pyrimidine-4,6-

( $5 \mathrm{H}, 7 \mathrm{H}$ )-diones (5a-d). General Procedure.-Method $A$. A solution of a hydrazone (4) ( 0.002 mol ) in dimethylformamide ( 20 ml ) was refluxed for 3 h . The mixture was evaporated to dryness in vacuo and the residue was recrystallized from ethanol to give the pyrazolo[3,4-d]pyrimidine (5) (Table 1). M.p.s and yields are indicated below the formulae.

To a solution of 1,3-dimethyl-6-hydrazinouracil (6 g, 0.035 mol ) in ethanol ( 180 ml ) was added crotonaldehyde ( $3 \mathrm{~g}, 0.042 \mathrm{~mol}$ ), and the mixture was stirred at room temperature for 1 h . The crystals which separated were filtered off, dried, and recrystallized from ethanol to give prisms ( $5.5 \mathrm{~g}, 70.5 \%$ ), m.p. $216^{\circ}, M^{+} 222$ (Found: C, 54.3; $\mathrm{H}, 6.45$; $\mathrm{N}, 25.4$. $\quad \mathrm{C}_{10} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires C, $54.05 ; \mathrm{H}, 6.35$; N, 25.2\%).

(1)

$$
\begin{aligned}
& \mathrm{a} ; \mathrm{R}=\mathrm{Ph} \\
& \mathrm{D} ; \mathrm{R}=\rho-\mathrm{ClC}_{6} \mathrm{H}_{4} \\
& \mathrm{c} ; \mathrm{R}=3,4-\mathrm{Cl}_{2} \mathrm{C}_{6} \mathrm{H}_{3} \\
& \mathrm{~d} ; \mathrm{R}=\rho-\mathrm{MeO}_{6} \cdot \mathrm{C}_{6} \mathrm{H}_{4} \\
& \mathrm{e} ; \mathrm{R}=\rho-\mathrm{Me}{ }_{2} \mathrm{~N} \cdot \mathrm{C}_{6} \mathrm{H}_{4} \\
& \mathrm{f} ; \mathrm{R}=\mathrm{MeCH}: \mathrm{CH} \\
& \mathrm{~g} ; \mathrm{R}=\mathrm{PhCH}: \mathrm{CH}
\end{aligned}
$$


(2)

$$
\begin{aligned}
& \text { a; } R^{\prime}=R^{2}=P h \quad\left(193^{\circ} \mathrm{C} ; 80 \%\right) \\
& \text { b; } R^{\prime}=R^{2}=\rho-\mathrm{ClC}_{6} \mathrm{H}_{4} \quad\left(181^{\circ} \mathrm{C} ; 87 \%\right) \\
& \text { c; } R^{\prime}=R^{2}=3,4-\mathrm{Cl}_{2} \mathrm{C}_{6} \mathrm{H}_{3} \quad\left(194^{\circ} \mathrm{C} ; 85 \%\right) \\
& d ; R^{1}=R^{2}=\rho-\mathrm{MeO} \cdot \mathrm{C}_{6} \mathrm{H}_{4} \quad\left(160^{\circ} \mathrm{C} ; 84 \%\right) \\
& \text { e; } R^{1}=R^{2}=\rho-\mathrm{Me}_{2} \mathrm{~N} \cdot \mathrm{C}_{6} \mathrm{H}_{L} \quad\left(258^{\circ} \mathrm{C} ; 58 \%\right) \\
& f ; R^{1}=\rho-\mathrm{MeO} \cdot \mathrm{C}_{6} \mathrm{H}_{4}, \mathrm{R}^{2}=\rho-\mathrm{ClC}_{6} \mathrm{H}_{4}\left(204^{\circ} \mathrm{C} ; 68 \%\right) \\
& g ; R^{\prime}=p-\mathrm{Me}_{2} \mathrm{~N} \cdot \stackrel{\mathrm{C}}{6}^{2} \mathrm{H}_{4}, \mathrm{R}^{2}=\mathrm{Ph}\left(150^{\circ} \mathrm{C} ; 62 \%\right) \\
& h ; R^{\prime}=\mathrm{MeCH}: \mathrm{CH}, \mathrm{R}^{2}=\mathrm{Ph} \quad\left(126^{\circ} \mathrm{C} ; 32 \%\right) \\
& \text { i; } R^{1}=\mathrm{MeCH}: \mathrm{CH}, \mathrm{R}^{2}=\rho-\mathrm{ClC}_{6} \mathrm{H}_{4}\left(152^{\circ} \mathrm{C} ; 50 \%\right) \\
& j ; R^{1}=\mathrm{MeCH}: \mathrm{CH}, \mathrm{R}^{2}=3,4-\mathrm{Cl}_{2} \mathrm{C}_{6} \mathrm{H}_{3}\left(194^{\circ} \mathrm{C} ; 53 \%\right) \\
& k ; R^{1}=\mathrm{PhCH}: C H, R^{2}=\mathrm{Ph} \quad\left(162^{\circ} \mathrm{C} ; 73 \%\right)
\end{aligned}
$$


(3)
$a ; R=P h$
b; $R=\rho-\mathrm{ClC}_{6} \mathrm{H}_{4}$
c; $R=p-\mathrm{MeO} \cdot \mathrm{C}_{6} \mathrm{H}_{6}$

(4)

(5)

(6)
a; $R^{1}=R^{2}=P h \quad\left(231^{\circ} \mathrm{C} ; 68 \%\right)$
b; $R^{1}=R^{2}=p-\mathrm{ClC}_{6} \mathrm{H}_{4} \quad\left(267^{\circ} \mathrm{C} ; 87 \%\right)$
c; $R^{\prime}=R^{2}=\rho-\mathrm{MeO} \cdot \mathrm{C}_{6} \mathrm{H}_{4} \quad\left(247^{\circ} \mathrm{C} ; 78 \%\right)$
d; $R^{\prime}=p-\mathrm{MeO} \cdot \mathrm{C}_{6} \mathrm{H}_{4}, R^{2}=p-\mathrm{ClC}_{6} \mathrm{H}_{4}\left(200^{\circ} \mathrm{C} ; 65\right.$

(7)

(8)

(9)


(10)

$$
\begin{aligned}
& a ; R=P h \\
& b ; R=p-\mathrm{ClC}_{6} \mathrm{H}_{4} \\
& c ; R=3,4-\mathrm{Cl}_{2} \mathrm{C}_{6} \mathrm{H}_{3} \\
& d ; R=p-\mathrm{MeO}^{2} \cdot \mathrm{C}_{6} \mathrm{H}_{4}
\end{aligned}
$$

Method B. A mixture of a 6-benzylidenehydrazino-3methyluracil (3) ( 0.008 mol ) and an aromatic aldehyde $(0.01 \mathrm{~mol})$ in dimethylformamide ( $60-80 \mathrm{ml}$ ) was refluxed for 7 h , then evaporated to dryness in vacuo, and the residue was recrystallized from ethanol to give (5) in almost the same yields as from method A.

Crotonaldehyde 1,3-Dimethyluracil-6-ylhydrazone (1f).-

Cinnamaldehyde 1,3-Dimethyluracil-6-ylhydrazone (1g).To a solution of 1,3-dimethyl-6-hydrazinouracil ( 2 g , 0.012 mol ) in ethanol ( 150 ml ) was added cinnamaldehyde ( $2.1 \mathrm{~g}, 0.016 \mathrm{~mol}$ ) ; treatment as above gave prisms ( 2.84 g , $85 \%$ ), m.p. $249^{\circ}, M^{+} 312$ (Found: C, 63.4; H, 5.55 ; N, 20.0. $\quad \mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $\mathrm{C}, 63.35$; $\mathrm{H}, 5.65 ; \mathrm{N}, 19.7 \%$ ). 5,7-Dimethyl-2-(1-methylbenzyl)-3-phenylpyrazolo[3,4-d]-
pyrimidine-4,6(5H,7H)-dione (7).-A mixture of acetophenone 1,3-dimethyluracil-6-ylhydrazone (6) ( $2 \mathrm{~g}, 0.0074$ mol ) and benzaldehyde ( $1.2 \mathrm{~g}, 0.01 \mathrm{~mol}$ ) in dimethylformamide ( 40 ml ) was refluxed for 6 h , then evaporated to dryness, and the residue was recrystallized from ethanol to give granules ( $1.55 \mathrm{~g}, 59 \%$ ), m.p. $148^{\circ}, M^{+} 360$ (Found: C, $69.9 ; \mathrm{H}, 5.65 ; \mathrm{N}, 15.3 . \quad \mathrm{C}_{21} \mathrm{H}_{20} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $\mathrm{C} 70.0 ; \mathrm{H}$, 5.6; N, $15.55 \%$ ).

6-Amino-3-aryl-2-benzyl-4-hydroxypyrazolo[3,4-d]pyrimidines (9a-d). General Procedure.-Method A. A mixture of 2 -amino-6-hydrazino-4-hydroxypyrimidine (10) $(0.01 \mathrm{~mol})$ and an aromatic aldehyde $(0.024 \mathrm{~mol})$ in dimethyl-
formamide ( 50 ml ) was refluxed for 5 h . After cooling, the crystals precipitated were filtered off and recrystallized from dimethylformamide [yields $60-90 \%$ (Table 2)], m.p. $<360^{\circ}$. The structures ( 9 ) were confirmed by the benzylic proton n.m.r. signals at $\delta 5.7-5.5$. (Table 3 in Supplementary Publication).

Method B. Compounds (9a-d) were also prepared by the condensation of 2 -amino-6-benzylidenehydrazino-4hydroxypyrimidines ( $8 \mathrm{a}-\mathrm{d}$ ) and aromatic aldehydes in dimethylformamide in similar yields.


[^0]:    $\dagger$ In the benzylation, the isomeric 1-benzylpyrazolo[3,4- $d$ ]-pyrimidine-4, $6(5 H, 7 H)$-diones were obtained as minor products.
    $\ddagger$ Tables 1-3, containing analytical and n.m.r. data, are available as Supplementary Publication No. SUP 21975 ( 4 pp.). For details of Supplementary Publications see Notice to Authors No. 7, J.C.S. Perkin I, 1976, Index issue.
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